



Investigation of the well-dispersed magnetorheological oil-based suspension with superparamagnetic nanoparticles using modified split Hopkinson pressure bar

Magdalena Osial^{1,2} · Michał Nowicki³ · Ewa Klejman¹ · Leszek Frańś²

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Abstract

Magnetorheological (MR) fluids are classified as smart materials whose viscoplastic characteristics change under the magnetic field. They are widely applied for dynamic energy dissipation due to their rapid thickening under the external magnetic field. In this work, the core–shell suspension of superparamagnetic iron oxide-based nanoparticles was synthesized and dispersed in silicone oil. Much effort has been made to prepare suspension meeting requirements of MR fluid. The experimental squeezing flow response was studied using a modified split Hopkinson pressure bar (SHPB) with various shear rates. Tests with modified SHPB show that MR fluid rapidly responds to the compression thickening and forming chain-like structures. MR fluid dissipates the energy generated during compression stress tests. This study presents a simple and cost-effective synthesis way suitable for MR fluid formation for its dynamic energy dissipation application.

Keywords SPION · Magnetorheological fluid · Split Hopkinson pressure bar · High strain rate · Dynamic behaviour

Abbreviations

| | |
|----------|---|
| A_I | Cross-section area |
| A_T | Cross-section area of the hollow |
| C_0 | Speed of a stress wave in the bar |
| D | Diameter of the bar |
| D_{S0} | Diameter of specimen |
| $e(t)$ | Axial strain |
| $e_I(t)$ | Incident wave |
| $e_T(t)$ | Reflected wave |
| $e_R(t)$ | Transmitted wave |
| E | Young modulus |
| FT-IR | Fourier transformed infrared spectroscopy |
| l_0 | Length of the specimen |
| MR | Magnetorheological |

| | |
|--------------------------|--|
| MRF | Magnetorheological fluid |
| $\dot{\epsilon}_{ns}(t)$ | Nominal strain rate |
| NP | Nanoparticle |
| OA | Oleylamine |
| $\sigma_{ns}(t)$ | Nominal strain |
| SEM | Scanning electron microscopy |
| SHPB | Split Hopkinson pressure bar |
| SPION | Superparamagnetic iron oxide nanoparticle |
| SPION@OA | Superparamagnetic iron oxide nanoparticle capped with oleylamine |
| TEM | Transmission electron microscopy |
| TGA | Thermogravimetry |

Introduction

Magnetorheological (MR) fluids are categorized as smart materials, and they are one of the most functional fluids changing rheological properties under the application of the external magnetic field (Acharya et al. 2020). Such field-responsive fluids can reverse their viscosity forming a chain-like structure (Ashour et al. 1996, Muhhamed et al. 2006), while without magnetic field MR fluid behaves like Newtonian (Bica 2006). Formation of the solid-like structures of fibril shapes due to the dipole–dipole interactions between particles under the adjusted magnetic field leads

Magdalena Osial and Leszek Frańś contributed equally to this work.

✉ Magdalena Osial
mosial@chem.uw.edu.pl

¹ Faculty of Chemistry, University of Warsaw, Pasteura 1, 02-093 Warsaw, Poland

² Institute of Fundamental Technological Research of the Polish Academy of Science, Pawińskiego 5B, 02-106 Warsaw, Poland

³ Institute of Metrology and Biomedical Engineering, Warsaw University of Technology, Boboli 8, 02-525 Warsaw, Poland

to the increased viscosity and strength of the fluid with stable shear resistance (Susan-Resiga et al. 2012, An 2017). Such unique magnetorheological properties of MR fluids open possibilities for their application in many fields, especially in mechanical devices like valves, dampers, clutch or brakes to reduce different vibrations (Kciuk and Turczyn 2006, Pręgoska et al. 2013, Ashtiani et al. 2015, Kumar and Kumar 2015). MR fluids are usually made of ferromagnetic colloidal suspension dispersed in liquid media like oil (Premalatha et al. 2012, Galindo-Gonzalez et al. 2016, Zhu et al. 2019). Although MR fluids reveal the ability to respond rapidly to the magnetic field, they tend to sediment decreasing their long-term stability (Wereley et al. 2006, Park et al. 2009). For that reason, much progress has been made towards improving the stability of MR fluids, including fluids differing from iron particles-based have been investigated (Seo et al. 2016, Guo et al. 2018).

Among various ferromagnetic materials, magnetic nanoparticles (NPs) based on iron oxide have shown high magnetization saturation and small hysteresis values, which makes them a promising material for MR fluids (Espin et al. 2005, Rabbani et al. 2019). Despite the wide application of the micron-sized particles in classical MR fluids, recent studies show the high potential of nano-sized particles in MR fluid preparation (Chae et al. 2015, Wu et al. 2016, Saha et al., 2019).

Within our studies, even very small particles obtained size up to 20 nm revealing superparamagnetic modified stabilized with organic molecules can be applied as the magnetorheological fluid.

In this work, MR fluid based on the core-shell superparamagnetic iron oxide nanoparticles (SPIONs) capped with oleylamine (OA) known as an effective stabilizing agent for various nanoparticles, including magnetic particles, was synthesized and investigated (Peng et al. 2006). The synthesis was performed with a simple and cost-effective coprecipitation technique (Gawęda et al. 2020, Rashid et al. 2020). The suspension was dispersed in silicone oil, and several methods were used for the investigation of the colloid properties. Additional experiments aimed at measuring the dynamic energy dispersion have been carried using split Hopkinson pressure bar (SHPB) (Lim et al. 2010a), being a technique widely used for testing the strain compression of various materials having industrial application (Shim and Mohr 2009, Lim et al. 2010b). Within SHPB measurements, the response of colloidal SPION@OA suspension under high strain rates (Fraś 2015) and determination of the time needed for the transition were investigated (Lim et al. 2010b). So far, our attention has been paid to the study of stress relaxation of our MR fluid under stress to provide information towards the description of the mechanical properties of our fluid. Additionally, the classical bar-bar system (Fraś and Pęcherski 2018) was improved and substituted

with a bar-hollow system, improving the signal detection for even minimal rheological changes of the colloidal under stress.

Experimental

Chemicals

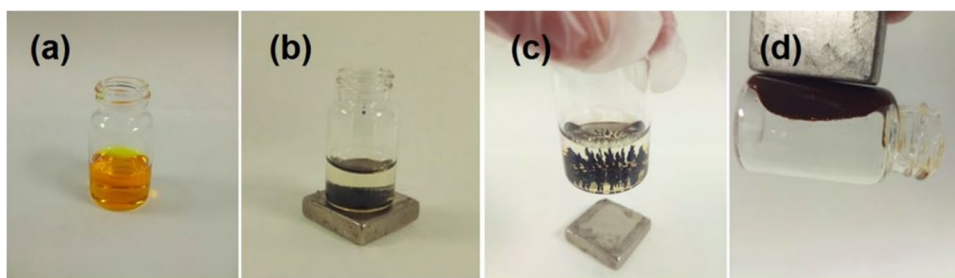
Iron (III) chloride hexahydrate $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ Aldrich ACS reagent 97% and iron (II) chloride tetrahydrate $\text{FeCl}_2 \cdot 4\text{H}_2\text{O}$ puriss p.a. $\geq 99\%$ (RT) were supplied from Sigma-Aldrich. Twenty-five percent ammonia solution NH_4OH was provided from POCH. Deionized water with resistivity 18.2 $\text{M}\Omega \text{ cm}$ at 25 °C was obtained using the Milli-Q ultra-pure water filtering system from Merck. Ethanol was purchased from Chempur with 96% analytical grade. Acetone was purchased from POCH with commercial grade. Synthesized SPIONs were modified with oleylamine puriss $\geq 98\%$ obtained from Sigma-Aldrich. The silicone oil for high temperatures was purchased from ACROS Organics.

Synthesis and modification of SPIONs

Iron oxide nanoparticles were synthesized with a coprecipitation method at room temperature from the solution containing 5.40 g of $\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$ and 1.99 g of $\text{FeCl}_2 \cdot 4\text{H}_2\text{O}$ in 100 mL of water in a glass beaker. Next, a 25% NH_4OH aqueous solution was added as a precipitation agent. Synthesis of iron oxide nanoparticles was carried out for 5 min at pH 10, with mechanical stirring 300 rpm. After synthesis, particles were separated from the solution on the magnet and rinsed several times with Milli-Q water until neutral pH. Next, the NPs were washed several times with ethanol and acetone, and 500 μL of oleylamine was added as a capping agent. Such compounds are widely used as an effective capping agent for the improvement of the stability of colloids (Lu et al. 2016). After OA addition, the suspension was mechanically mixed with 500 rpm for 30 min and then placed in the ultrasonic bath for 10 min to make the mixture evenly. After OA capping nanoparticles were collected on a magnet and ethanol was removed.

Two different kinds of volume fractions (65% and 75% OA capped particles content vs oil fraction) were prepared by mechanical mixing with 500 rpm for 60 min of SPION@OA with silicone oil at room temperature. Then, obtained suspension was completely uniform, and no additives were needed. Figure 1 shows the synthesis from (a) solution containing a source of iron (II) and (III) ions, (b) nanoparticles collected on a magnet after ammonia solution addition and stirring, (c) nanoparticles orienting along the direction of the magnetic field in water and (d) SPION@OA in oil in the presence of the magnetic field.

Fig. 1 Preparation of MR fluid a solution with a source of Fe(II) and Fe(III) ions before ammonia solution addition, **b** nanoparticles collected on the magnet in water, **c** MR fluid behaviour in the presence of the magnetic field, and **d** the MR fluid attracted by magnet



Methods

The size and shape of synthesized SPIONs were studied with transmission electron microscopy (TEM)—EF-TEM, Zeiss Libra 120 Plus, Stuttgart, Germany, operating at 120 kV, and scanning electron microscopy (SEM)—Merlin, ZEISS, Stuttgart, Germany.

The presence of organic compounds on the SPIONs surface was investigated by FT-IR spectroscopy with Nicolet 8700 Spectrometer (Thermo Scientific).

Thermogravimetric analysis (TGA) was performed with TGA Q50 (TA Instruments), New Castle, USA, under a nitrogen atmosphere.

The magnetic behaviour of samples was investigated with QD vibrating sample magnetometer (VSM) over the magnetic field range from -2.0 to $+2.0$ T in the 100 K and 300 K stabilized to the accuracy of about 0.01 K.

The microscopic images of MR fluid were examined through Olympus optical microscope BX 41-TF. The images were taken with the Olympus digital camera. The rheological properties were studied with and without a magnetic field using a parallel plate type system at room temperature, with the sinusoidal change of current.

Viscosity and shear stress curves of the investigated samples were measured on the specially constructed test stand, which main component was heavily modified rotary rheometer (Rheotest 2.1, Germany). Its torque meter was equipped with high precision ADC, and data acquisition software was written in LabVIEW. Furthermore, it was calibrated with 30 Pa*s silicone viscosity standard (AMETEK Brookfield, Germany) obtaining ± 0.1 Pa*s measurement uncertainty. Axial magnetic field was provided by either current

controlled coil (up to 50 mT) or specially designed ring permanent magnets (100 and 150 mT). Magnetic field was measured with teslameter (Lakeshore 475, USA).

The dynamic behaviour of samples was tested with its laboratory stand of split Hopkinson pressure bar setup. SHPB was made with 7075 alloy aluminium bar 1000×20 mm as an incident bar and hollow as a transmitted bar with the wall thickness 2 mm (see Fig. 2). The incident bar is working as a quarter bridge, but a transmitted bar is connected with a full-bridge sensor system. The signal from strain gauges is amplified by 400 V/V. The laboratory test stand is using the coil, with inner diameter 30 mm and length 50 mm having 80 turns of the wire with the final resistance 180 Ω . The power line was supplied with EA-PS 8360–30, where maximal direct voltage value is 360 V and the current is 2 A and it can generate 150 mT magnetic field, what is more, the SHPS is modified with the hollow (11 at Fig. 2).

The application of bar-hollow system instead of the classical bar-bar system improves mechanically recorded signals, allowing for a determination of even minimal deformations of tested colloidal suspension. The applied strain rate varied from $4.42 \cdot 10^3$ to $1.18 \cdot 10^4$ s $^{-1}$.

Results

Morphology, magnetic and spectroscopic properties

The shape and size of bare SPIONs and SPION@OA was investigated with transmission electron microscopy. As can be seen at Fig. 3a, SPIONs have a well-defined

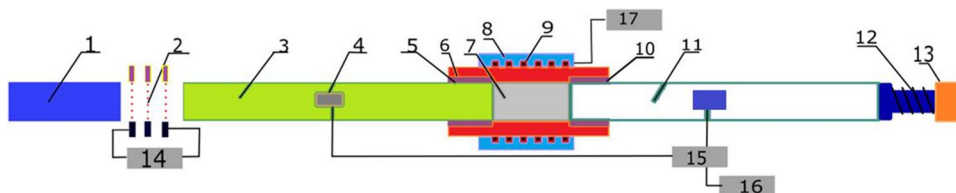
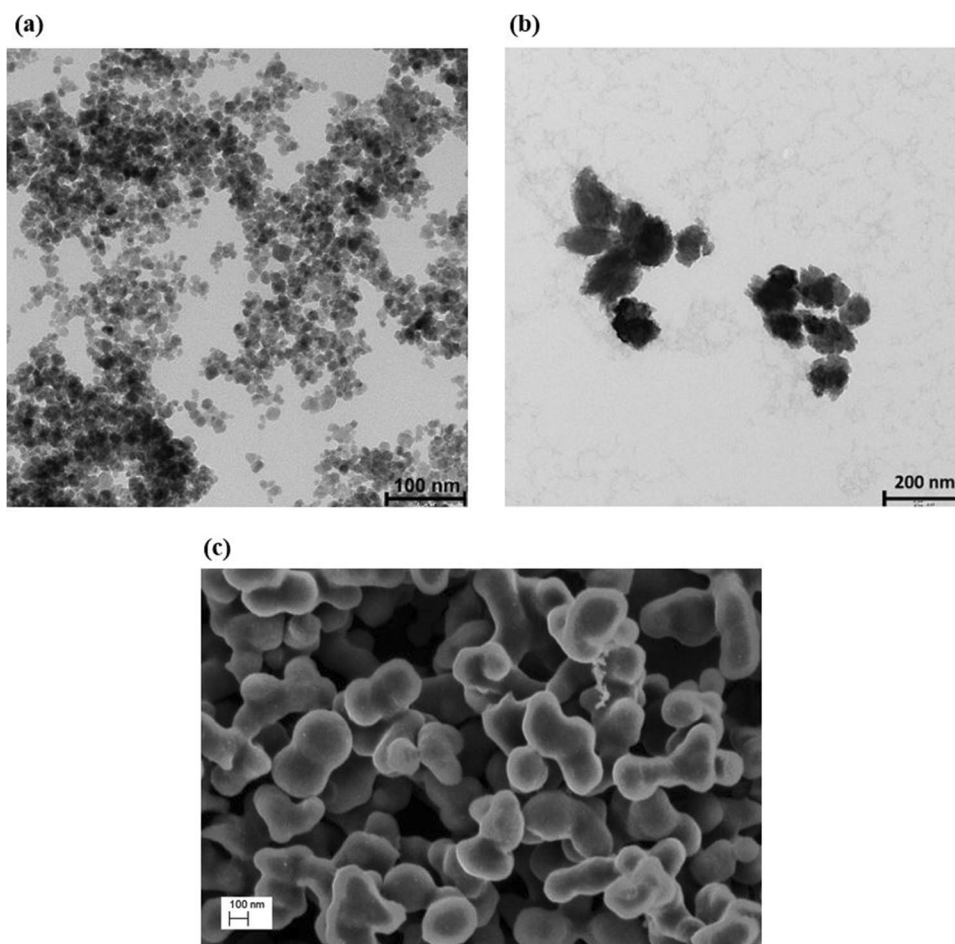


Fig. 2 Scheme of Hopkinson bar used for dynamic evaluation of synthesized MR fluid. (1) Striker, (2) laser sensor to measure velocity, (3) incident bar, (4) strain gauges, (5) sealing, (6) deformable hose,

(7) MR fluid, (8) coil cover, (9) coil, (10) sealing, (11) transmitter bar—hollow, (12) spring, (13) momentum trap, (14) velocity measure system, (15) signal amplifier, (16) oscilloscope

Fig. 3 **a** TEM images of bare SPION before surface modification, **b** TEM image of SPIONs capped with oleylamine, **c** SEM image of SPIONs capped with oleylamine



spherical shape with a measured average length axis of 10–15 nm, while after oleylamine capping nanoparticles agglomerate forming 100–130 nm diameter spheres with similar shape and size (see Fig. 4b). Additionally, scanning

electron microscopy was used for the determination of the shape and size of SPION@OA (see Fig. 4c).

The spectroscopic analysis shows that the magnetic core is effectively covered with oleylamine shell. As can be seen

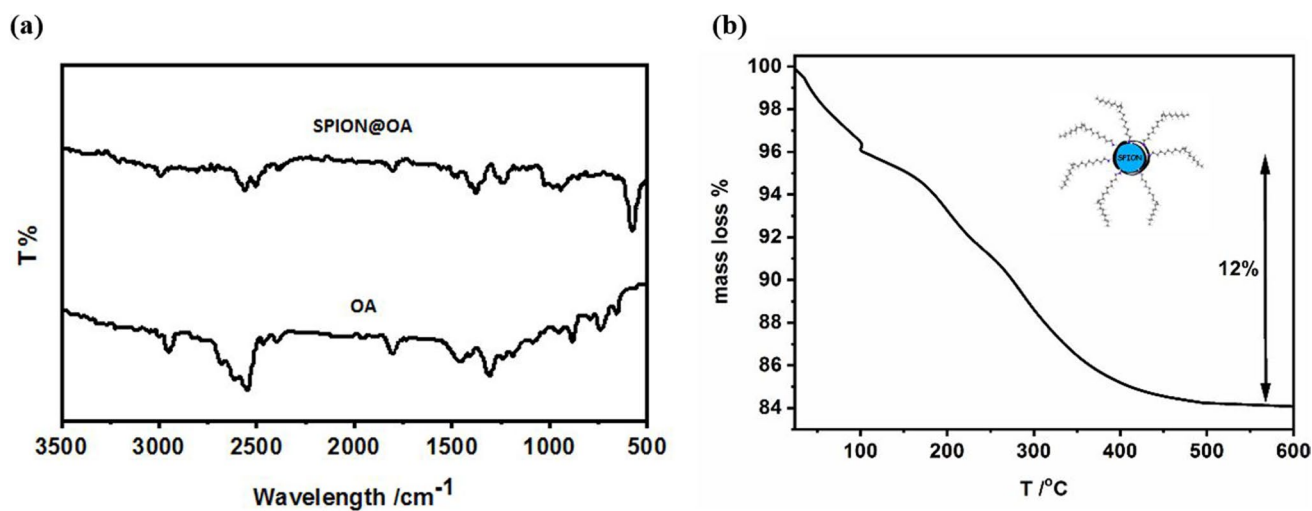


Fig. 4 **a** FT-IR spectra of SPION@OA and OA and **b** TGA curve for SPION@OA

at Fig. 4a, FT-IR spectra recorded for pure oleylamine and SPIONs capped with OA are similar, confirming that SPIONs are successfully covered with OA. Recorded spectra are in agreement with the literature data (Zhang et al. 2007, Kuman et al. 2014).

The band recorded at 590 cm^{-1} is characteristic for stretching Fe–O vibration at iron oxide particles. The signal around 1610 cm^{-1} is assigned to the N–H bending and NH_2 scissoring motion. For SPION@OA sample, the band at that region is much smaller than for OA measured separately indicating adsorption of OA on the surface of SPIONs. Both spectra reveal characteristic bands of terminal methyl asymmetric in-plane C–H stretching at 2900 cm^{-1} and asymmetric C–H stretching at 2850 cm^{-1} . The shift of bands for the SPION@OA sample may correspond to the constraint of the molecular motions due to the formation of close-packed layers of oleylamine on the SPIONs (Wang et al. 2014).

The thermogravimetry analysis was performed in the temperature range from ambient to $600\text{ }^\circ\text{C}$, with a heating rate of $10\text{ }^\circ\text{C}/\text{min}$. Measurement was performed under the nitrogen atmosphere (see Fig. 4b). The initial mass loss below $100\text{ }^\circ\text{C}$ corresponds to the elimination of water molecules adsorbed on the surface of SPIONs@OA. The following a drop of the mass is ascribed to degradation of the oleylamine degradation indicating 12% of OA on the surface of SPIONs.

The magnetic property of prepared material is a significant index influencing the magnetorheological property of MR fluid. Studies of as-synthesized iron oxide nanoparticles with magnetometry reveal the superparamagnetic behaviour of samples (see Fig. 5). The saturation magnetization of SPIONs was about 91 emu/g at 100 K and 82 emu/g at 300 K with low coercivity, which corresponds to data for SPIONs prepared in different conditions (Osial et al. 2018).

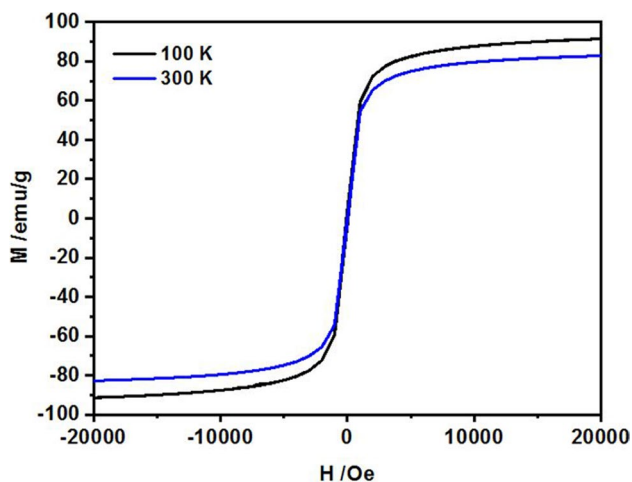


Fig. 5 Magnetization isotherms for SPIONs measured at 100 K and 300 K

Stability of MR fluid

The stability of MR suspension depends on the content of iron oxide nanoparticles in the suspension. In general, the sedimentation of particles dispersed in the liquid is not desirable property, especially in the case of MR fluid application for dynamic energy dissipation. Sedimentation tests were performed on two samples to compare the stability (see Fig. 6), and the sedimentation rate was calculated using Eq. 1:

$$\text{Sedimentation percentage}(S) = \frac{\text{Volume of supernatant fluid}}{\text{Total volume of the fluid}} \times 100\% \quad (1)$$

Sedimentation can be caused by the density mismatch between SPIONs@OA and oil, although the stability is much higher in comparison to other liquids, where the deposition reaches even 60% (Zhang et al. 2009, Jinaga et al. 2019, Kumar et al. 2019).

The suspension of SPIONs@OA in silicone oil was also investigated with an optical microscope showing homogeneous dispersion without and under a magnetic field (see Fig. 7). The experiments were performed on a glass slide, and the optical micrographs were taken in transmission mode. The applied magnetic field was about 10 mT . The direction of the current was changed with a sinusoid direction and the suspension structure was measured in a droplet sample having volume $10\text{ }\mu\text{L}$ diluted twice with oil was subjected to the magnetic field of strength using dual-mode coil. Initially, without the magnetic field (Fig. 7a), the nanoparticles in the suspension are dispersed randomly on the microscopic glass. When the magnetic field is induced, the NPs spontaneously start to form chain-like structures revealing MR properties (Fig. 7b–h) (Wang et al. 2020). Suspension quickly reorganized into

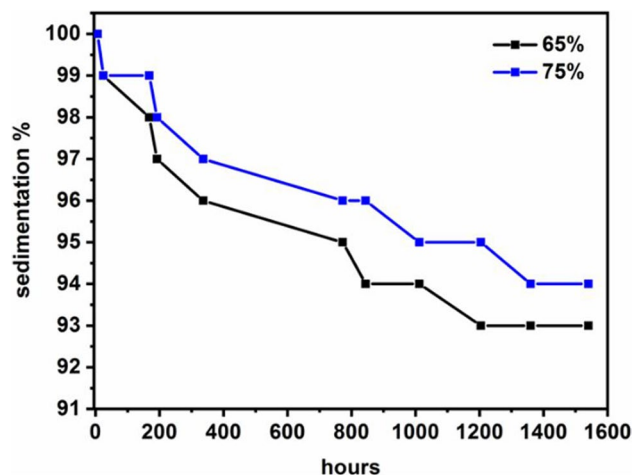
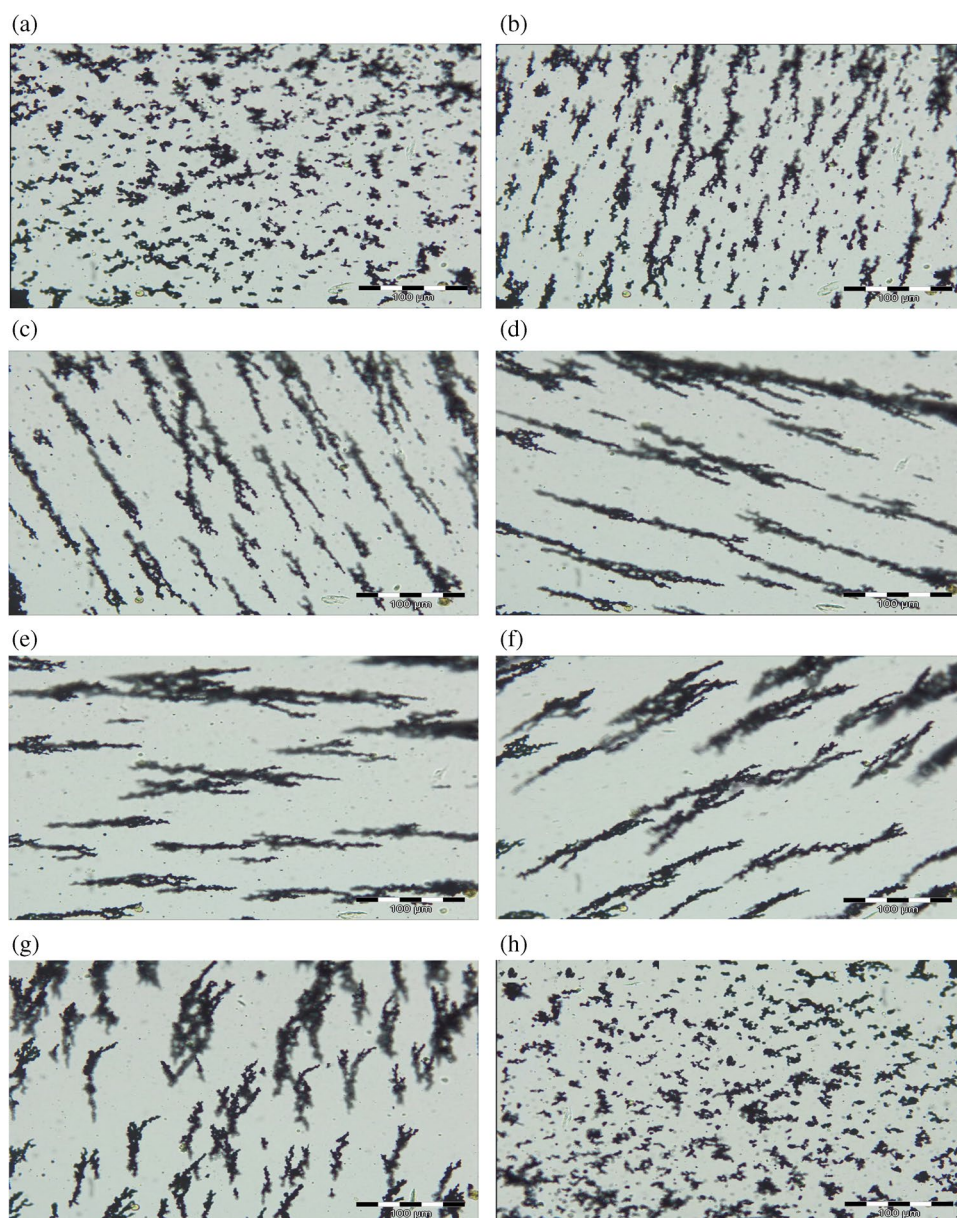


Fig. 6 Sedimentation of synthesized MR fluid in the function of time

Fig. 7 The silicon oil-based suspension (65% volume ratio) **a** without a magnetic field, **b–g** under magnetic field changing direction, and **h** without a magnetic field, with the scale bar of 100 μm



particle-rich zones orienting along with the direction of the magnetic field.

Application of the magnetic field causes agglomeration of the particles, which form the parallel zones. They rotate following the direction of the magnetic field forming chain-like structures, while when the magnetic field is off, the agglomerates disperse like before its application.

Viscosity studies

Additionally, the viscosity of as-synthesized MR fluid was tested within the concentric cylinder rotational viscometer, using 100 mL of MR fluid. Measurements were performed in the room temperature and without the magnetic field. As can be seen in Fig. 8, the viscosity of the as-synthesized MR

fluid decreases with the growth of the shear rate. Viscosity of MRF for higher content (about 75%) of solid nanoparticles is about 2 Pa*s higher than for lesser one (about 65% NPs).

Dynamic behaviour

Following studies of colloidal suspensions with 65% and 75% SPION@OA content in silicone oil were focused on the investigation of their viscoelastic behaviour. The gelatin was sealed to the bars by cyanoacrylate adhesive seal and taped too. The specimen length is 2 mm, and its diameter is 20 mm. Our fluid was placed in the thin gelatin hose (with wall thickness max. 1 mm) allowing deformation and observation of the sample behaviour during experiments.

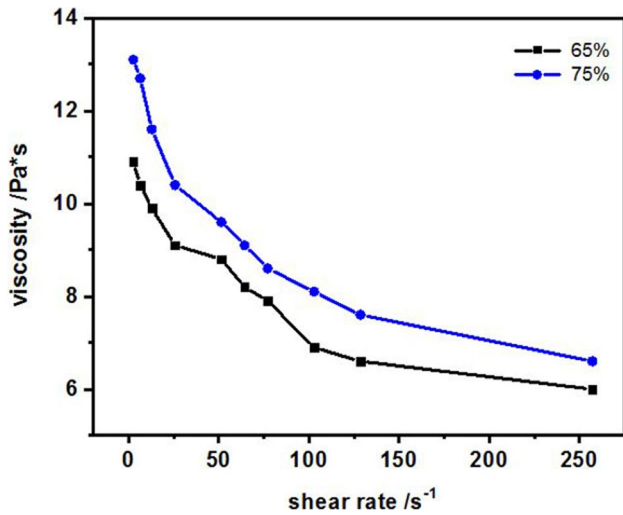


Fig. 8 The viscosity of MRF in function of shear rate

The dependence of the rearrangement of the microstructure vs applied stress was measured within a compression test. The results were obtained in which the nominal stress was permitted to vary from 0.001 to 200 MPa in a controlled manner over the duration of 0.5 ms. Both samples responded quickly occurring solid state under the magnetic field and liquid state without the magnetic field.

When the striker (1, Fig. 2) impacts the incident bar, then the generated stress wave is transmitted through the specimen and reaches the hollow. With the modification of the SHPR setup from the bar-bar system into the bar-hollow, it is possible to measure even very deformations of the tested fluid sample. Such distortions are recorded as the changes of the reflected wave (see Fig. 9). In our case, it was possible to record the wave for much smaller velocities of the striker than the literature shows (Wang et al. 2016).

The measured three waves—incident, reflected and transmitted—in the form of the axial strain $e(t)$ are described with the fundamental relation (see Eq. 2) making possible estimation of the stress and displacement. Firstly, the specimen during the test must be in the stress equilibrium.

$$E(e_I + e_R)A_I = Ee_TA_T, \tag{2}$$

where E is the Young modulus of the incident and transmitted bars. The symbol A_I denotes the cross-section areas of the incident bar in the place of glued strain gauges and A_T is the cross-section area of the hollow. To calculate the nominal strain, the following equations were used (Chen et al. 1999):

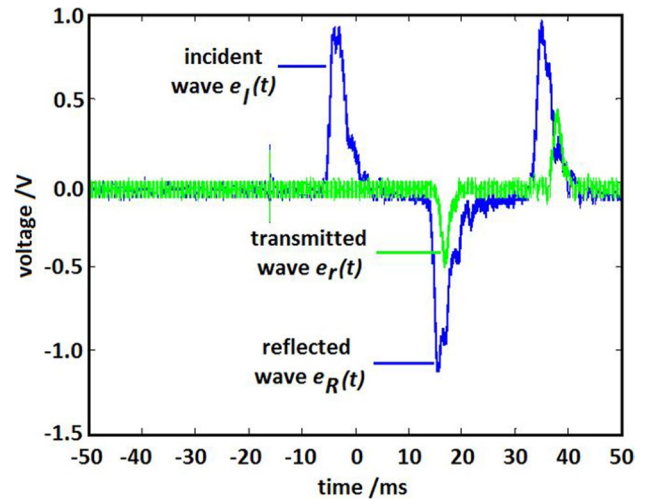


Fig. 9 The stress waves recorded by an oscilloscope

$$\epsilon_{ns}(t) = \frac{C_0}{l_0} \left(1 - \frac{A_I}{A_T}\right) \int_0^t e_I(t)dt + \frac{C_0}{l_0} \left(1 + \frac{A_I}{A_T}\right) \int_0^t e_R(t)dt, \tag{3}$$

where C_0 is the speed of a stress wave in the bar and l_0 is the length of the specimen.

$$\sigma_{ns}(t) = E \left(\frac{D}{D_{S0}}\right)^2 \left[\left(1 + \frac{A_T}{A_I}\right) e_T(t) \right], \tag{4}$$

where D and D_{S0} are ascribed to the diameter of the bar and specimen. Equation 5 presents the relation to calculate the nominal strain rate:

$$\dot{\epsilon}_{ns}(t) = \frac{1}{l_0} \left[\frac{du_a(t)}{dt} - \frac{du_b(t)}{dt} \right] = \frac{1}{l_0} \frac{d\epsilon_{ns}(t)}{dt} \tag{5}$$

In the beginning, when the striker hits the incident bar, the stress wave is propagating with the speed C_0 through the incident bar and compressing the material in gelatin hose. A slow increase as the nominal compressive stress occurs. The rise of the material response corresponds to the interaction of the fluid movement with the magnetic particles. Then, the MR fluid hardens, and the single peak is observed ascribed as the formation of the single chains in the fluid. After reaching the elastic limit, the curve drops, which corresponds to the dynamic recovery of the tested MR fluid. The rate of the deformation of chain-like structures changes with the speed of striker—which was changed from 8 to 16 m/s with the magnetic field 150 mT measured in the middle of the coil.

Maximum observed on the following curves corresponds to the steady-state regime with chain-like structures, while behind such region the curve drops slowly reaching the flow state. Measurements confirm that fluid

material hardens under the stress wave application and flows after the relaxation.

As can be seen in Figs. 10–11, the properties of the examined fluid depend on deformation velocities. The increase of deformation speed affects the yield point increasing its value. The magnetorheological fluid responds to this stress and the characteristic wave appears due to the deformation.

As-synthesized MR fluid based on iron oxide nanoparticles reacts to sudden stress and disperses the elastic longitudinal waves. The viscoelastic properties of these functional materials are based on the structure ready to rearrangement. The deformation chain-like structure is mostly based on shifting a single iron particle and swaps them into the neighbouring chain structure. This mechanism provides the deformation of the structure. The stress values obtained for material with 65% of volume ratio are much lower than the material with 75%. As can be seen in Fig. 10, the behaviour of MR fluid is dependent on the amount of magnetic nanoparticles that are responsible for viscoelastic deformation.

The best results in energy dispersion are obtained for MR fluid having 75% volume ratio for the specimen with yield

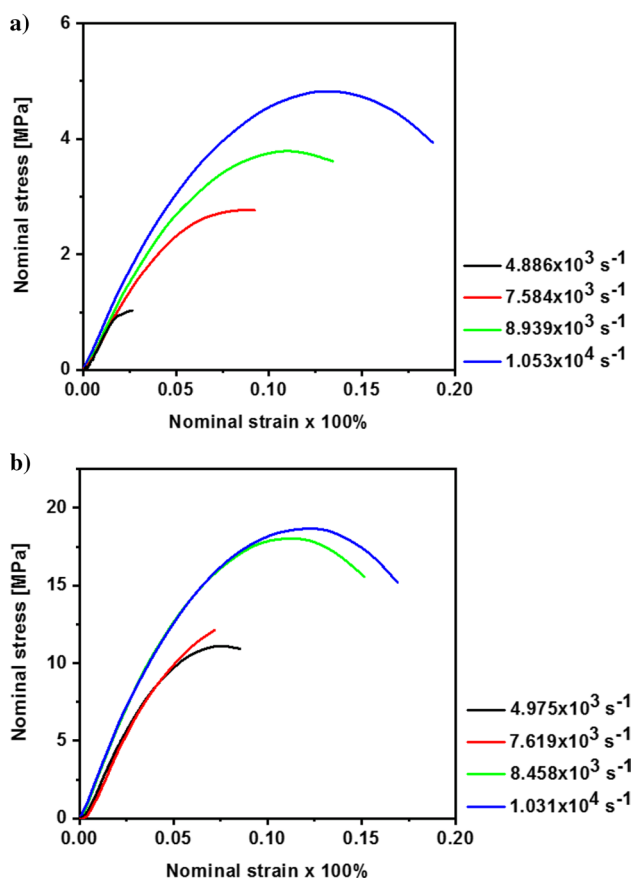


Fig. 10 Yielding for MR fluids having **a** 65% volume ratio and **b** 75% volume ratio. Measurement was performed under the 50 mT magnetic field

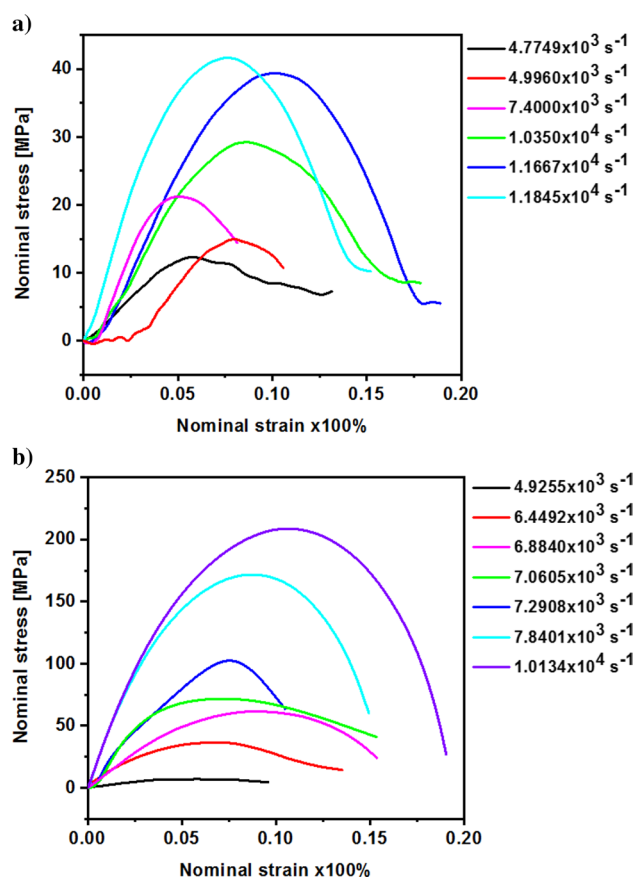


Fig. 11 Yielding for MR fluids having **a** 65% volume ratio and **b** 75% volume ratio. Measurement was performed under the 150 mT magnetic field

stress (the straight line crossing the stress–strain relation shifted by 0.2% relative to the elastic range) $1.2434 \cdot 10^8$ Pa than MR fluid having 65% volume ratio, which reaches the yield stress $2.6883 \cdot 10^7$ Pa. The strain rate of the specimen is closely similar, and some inaccuracies can come from another velocity of the strikers (15.84 vs 16.04 m/s).

Our studies show that the effect of SPION@OA volume fraction on the mechanical response is observed. Results obtained for MR fluid with a 65% volume ratio are comparable with literature for hard-sphere silica particles colloidal suspensions. In comparison, a 10% rise of volume ratio improves results for even one order of magnitude (Lim et al. 2010b). In general, MR fluid is a material that has a dipole–dipole structure (Jolly et al. 1996), so in presence of a magnetic field, these structures form a chain-like structure that is far from the initial one. While the incident wave transits through the sample, the structure undergoes deformation of the structure formed by particles and the sample has no time for quick relaxation.

Additionally, commercially available MR fluid (MRF) produced by the LORD Corporation—the MRF 140-CG that contains 85.44% of ferroelements was measured with

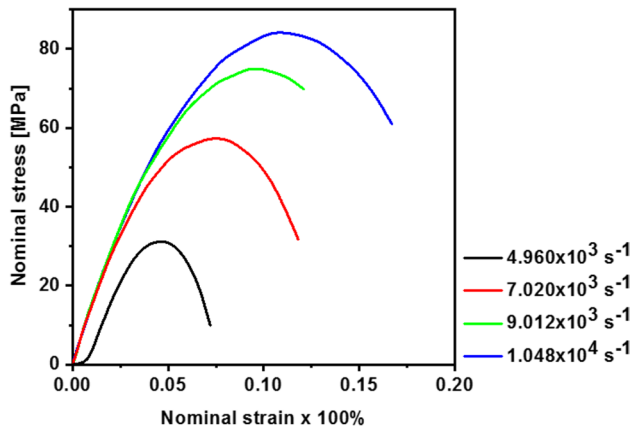


Fig. 12 Yielding for MRF 140-CG that contains 85.44% volume ratio. Measurement was performed under the 50 mT magnetic field

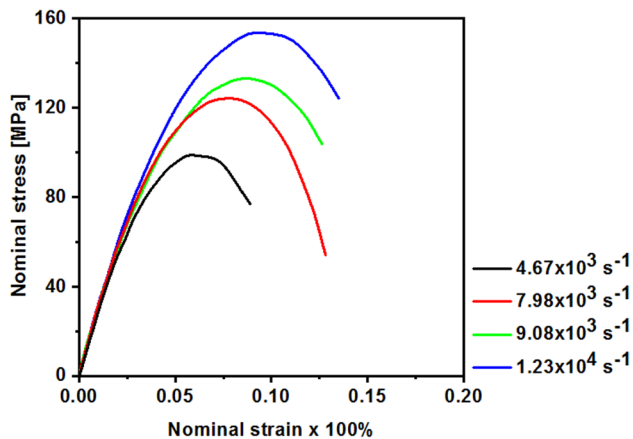


Fig. 13 Yielding for MRF 140-CG that contains 85.44% volume ratio. Measurement was performed under the 150 mT magnetic field

the use of SHPB. Likewise, the MRF that was synthesized in this project, various striker velocities were applied. Therefore, the generated by the impact compression wave, measured by the strain gauges, allowed to calculate the nominal stress, nominal strain and strain rate. The fluid responses to this stress and the characteristic wave that appears due to the deformation are similar to the results obtained for previous samples.

Figure 12 presents an experiment performed at the magnetic field about 50 mT, while Fig. 13 shows data obtained under 150 mT.

The results confirm that MRF material solidifies similarly, where ferroelements that form chain-like structures are responsible for effective dynamic energy dissipation. According to the data obtained for commercial MRF presented in Fig. 12 and 13, the as-synthesized sample behaves slightly different than the commercial one when the velocity of the striker is low, while for the large values

of velocity the efficiency is similar to the commercial MRF LORD.

The as-synthesized sample having 65% volume-ratio content has less chain-like structures, while 75% more respectively, while its susceptibility against deformation depends on the internal interactions between particles as well as the magnetic field. Under a weak magnetic field, particles form the less chain-like structure that influences the response of the sample, and as a result, lower values of stress–strain are measured. In comparison to the data that were recorded for commercially available MRF, results obtained with SHPB equipment for as-synthesized sample confirm that material can be used successfully in vibration dissipation which may have an adverse influence on mechanical devices and organisms.

Complementary to the SHPB tests, the dynamic viscosity changes and shear stress in the presence and absence of the magnetic field were investigated. Changes in the viscosity in the function of the shear rate demonstrate a non-Newtonian character for tested MR fluid. Here, the 75% colloidal suspension was tested. An increase of the viscosity is observed, where the values rise about 90 Pa*s when the magnetic field rises from 0 to 25 mT due to the formation of the chain-like structures in a presence of magnetic field. Significant change is also observed for the following rise of a magnetic field from 25 to 50 mT, while in 100 mT and 150 mT the viscosity remains similar (see Fig. 14a). These values decrease with an increase of the shear rate, which shows behaviour typical for pseudoplastic fluids with linearization of the characteristic above 3 s^{-1} shear rate and is typical for fluids with suspended particles. On the other hand, Fig. 14b shows the decrease of the viscosity to half of the initial value with the rise of the temperature, and with the rise of the shear rate above the value 5 s^{-1} it behaves similarly despite the temperature changes. Next, the shear stress was investigated in the function of shear rate. As can be seen in Fig. 14c, the changes in the shear stress values over the controlled shear rate mode increase with the increase in magnetic fields that is characteristic of the shear-thickening of the MR fluid. Following Fig. 14d presents the change of the relative viscosity in the function of shear stress. As shown below, these characteristics are monotonous for smaller rate of a magnetic field, decreasing with increasing shear rate, while for higher fields they have a maximum between 1 and 3 s^{-1} .

Conclusions

In summary, highly magnetic single crystallites based on the iron oxide nanoparticles obtained in this work were synthesized by fast, simple and cost-effective co-precipitation technique and easily modified with oleylamine as the stabilizing agent. TEM studies reveal spherical shape with an

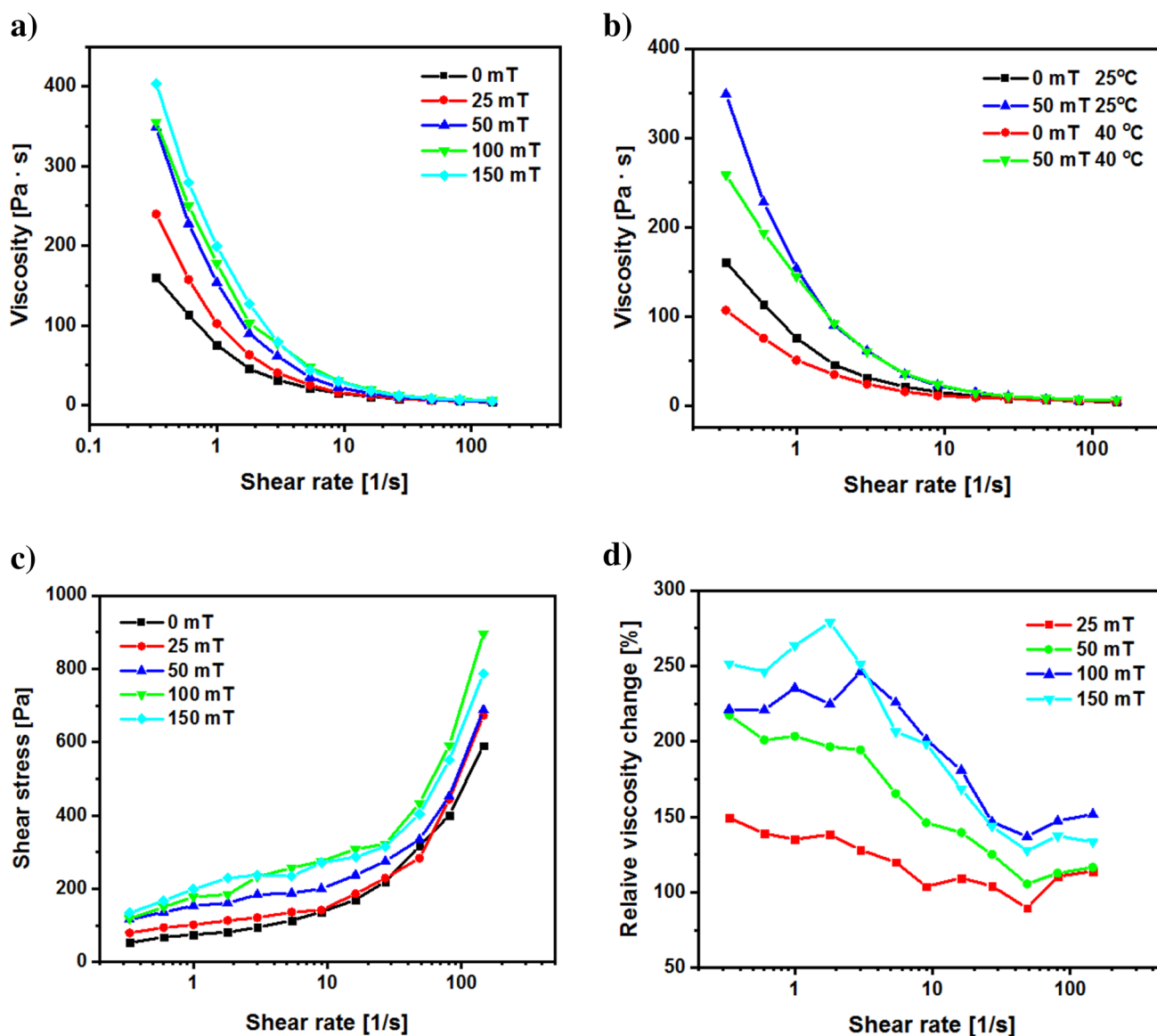


Fig. 14 Viscosity change in function of shear rate at **a** room temperature, **b** 40 °C and room temperature, **c** shear stress vs shear rate and **d** relative viscosity change vs shear rate. All measurements were performed in function of magnetic field

average size of about 10–15 nm before OA capping, while after SPION@OA creates homogeneous suspension with 100 nm size and spherical shape. Complex characterization confirms the successful capping of the magnetic core. Thermogravimetry analysis reveals that organic shell constitutes ca. 10% for the core of the conjugate mass. In the following studies, SPION@OA suspension was dispersed in the silicon oil, and two different MF fluids containing 65% and 75% volume ratios were tested. An increase of concentration of capped nanoparticles in the MR fluid improves the stability of the fluid and performance against applied stress.

Split Hopkinson pressure bar modification from bar-bar into the bar-hollow system enabled to record of the transmitted wave during energy dissipation experiments with higher

precision making it possible to record even tiny waves generated in the investigated material. Fluid prepared within this work responds reversibly and instantly to the stress occurring high yield stress, which makes it applicable in vibration reduction. Viscoelastic properties of 75% MRF for high striker velocities are comparable with commercial LORD MRF-140CG containing 85.44% of magnetic particles in the fluid. MR fluid having a 75% volume ratio reveals a promising effect of dynamic energy dispersion, giving the opportunity to test it in brakes and dampers.

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Declarations

Conflict of interest Authors declare no conflict of interest. The founding has no role in the design of the study and in the collection, analyses, data interpretation, manuscript writing and decision to publish the results.

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